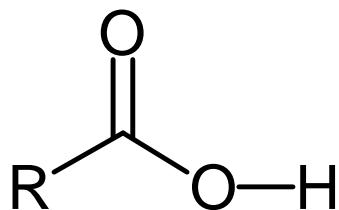


# 第十八章 羧酸及其衍生物 (carboxylic acids and their derivatives)



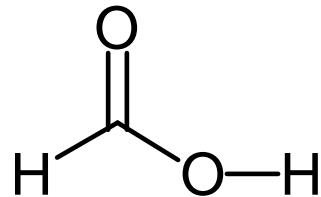
Structure	Name	Structure	Name
$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{Cl} \end{array}$	Acyl (or acid) chloride	$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{NH}_2 \end{array}$	Amide
$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{O}-\text{C}-\text{R}' \end{array}$	Acid anhydride	$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{NHR}' \end{array}$	
$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{O}-\text{R}' \end{array}$	Ester	$\begin{array}{c} \text{O} \\ \parallel \\ \text{R}-\text{C}-\text{NR}'\text{R}'' \end{array}$	
$\text{R}-\text{C}\equiv\text{N}$	Nitrile		

# 1) 命名

a) Acid:

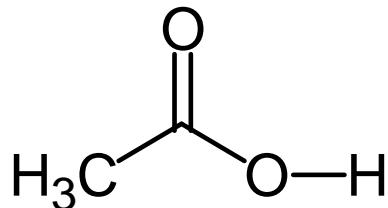
Systematic name

Common name



Formic acid

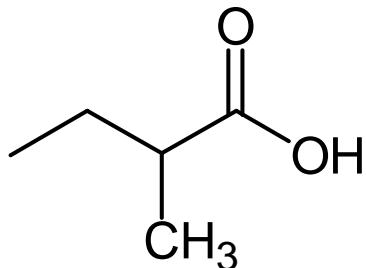
methanoic acid



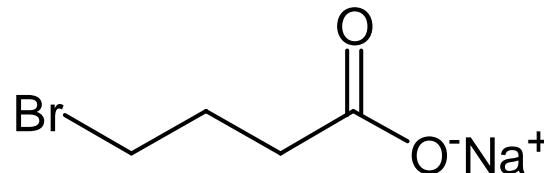
ethanoic acid

Acetic acid

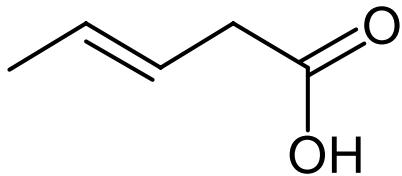
Exercise 816:



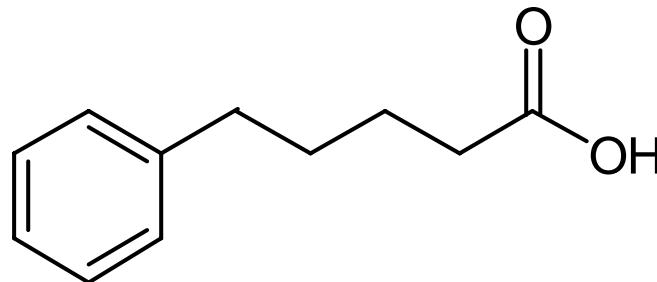
2-methylbutanoic acid



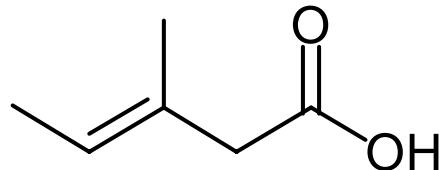
sodium 4-bromobutanate



(E)-3-pentenoic acid



5-phenylpentanoic acid

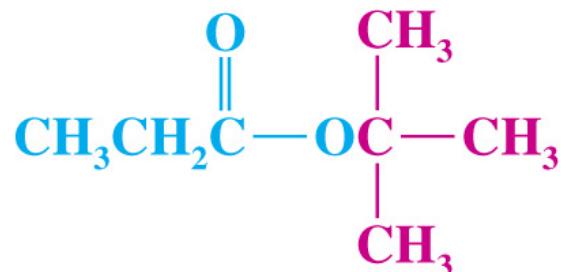


(E)-3-methyl-3-pentanoic acid

b) Esters:



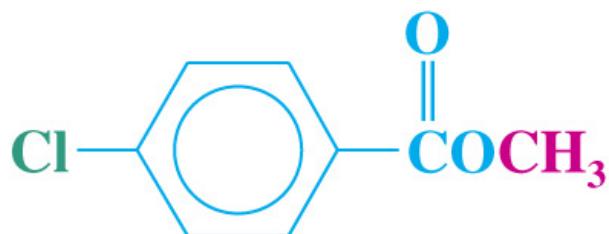
Ethyl acetate or  
ethyl ethanoate



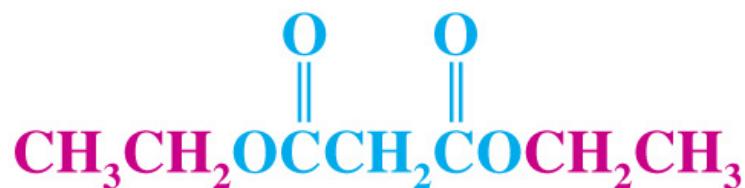
*tert*-Butyl propanoate



Vinyl acetate or  
ethenyl ethanoate

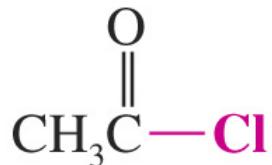


Methyl *p*-chlorobenzoate

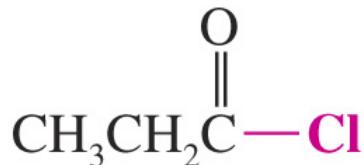


Diethyl malonate

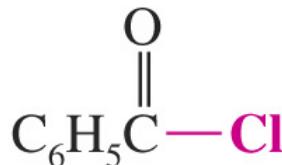
c) Acyl chlorides



**Acetyl chloride**  
**(ethanoyl chloride)**  
mp  $-112^\circ\text{C}$ ; bp  $51^\circ\text{C}$

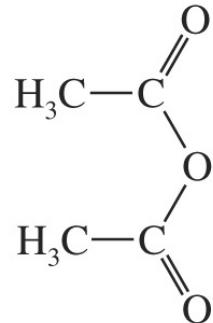


**Propanoyl chloride**  
mp  $-94^\circ\text{C}$ ; bp  $80^\circ\text{C}$



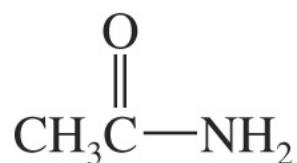
**Benzoyl chloride**  
mp  $-1^\circ\text{C}$ ; bp  $197^\circ\text{C}$

d) Carboxylic anhydride

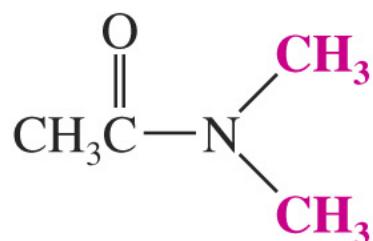


**Acetic anhydride**

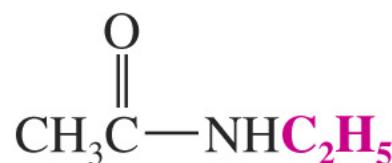
## e) Amides



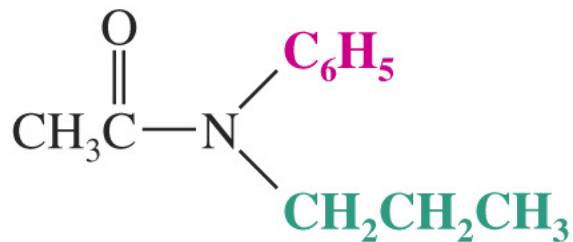
**Acetamide**  
(ethanamide)  
mp 82°C; bp 221°C



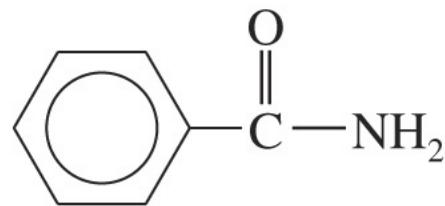
***N,N*-Dimethylacetamide**  
mp -20°C; bp 166°C



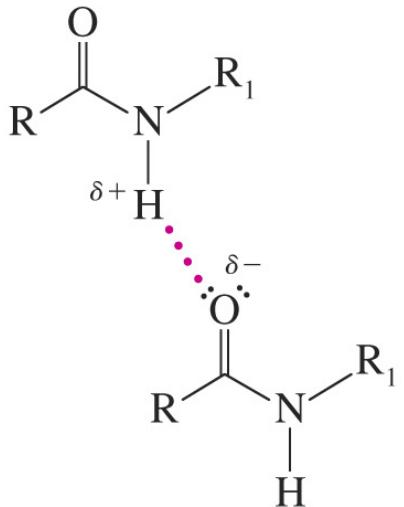
***N*-Ethylacetamide**  
bp 205°C



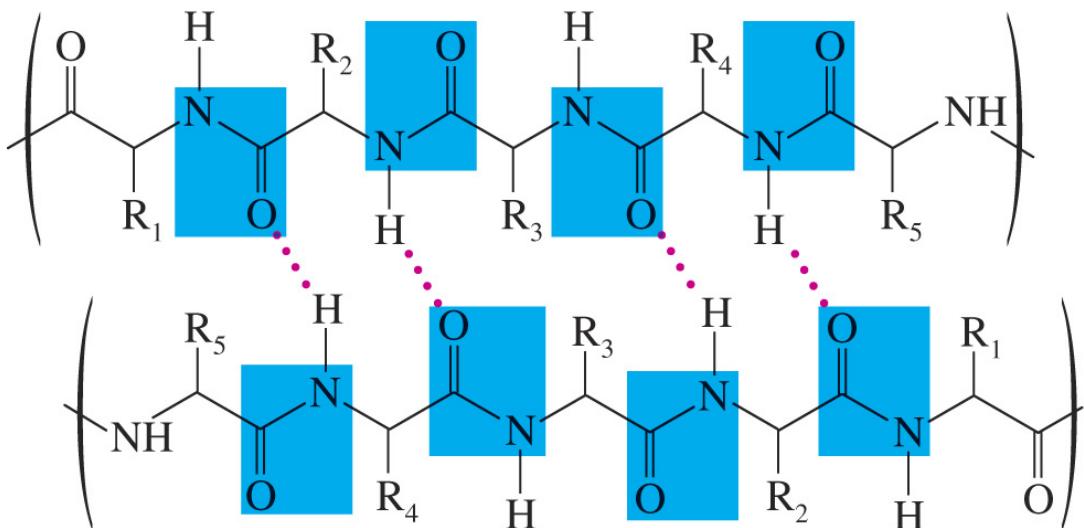
***N*-Phenyl-*N*-propylacetamide**  
mp 49°C; bp 266°C at 712 torr



**Benzamide**  
mp 130°C; bp 290°C

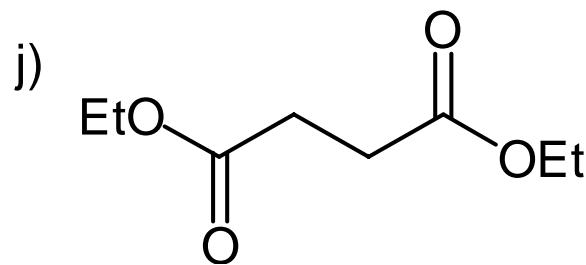
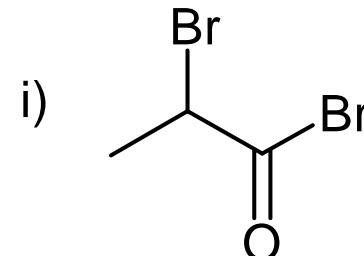
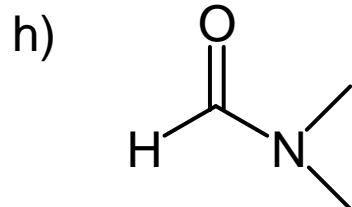
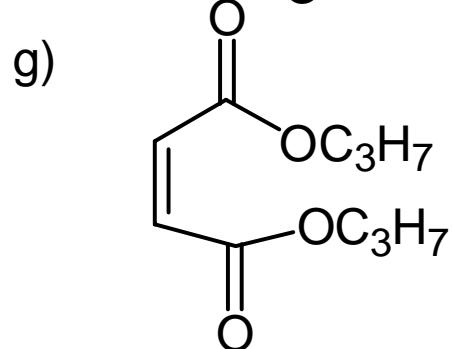
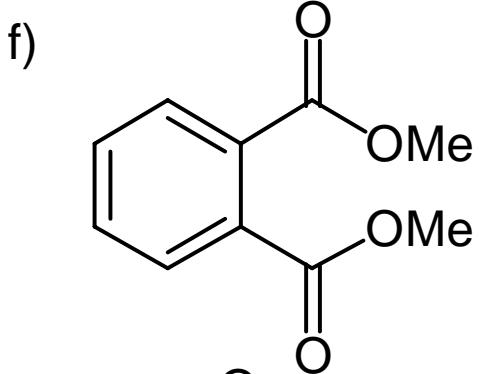
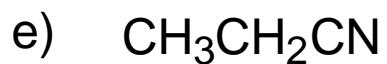
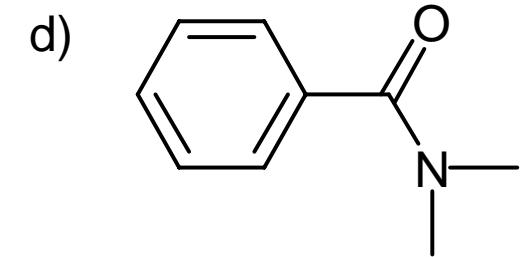
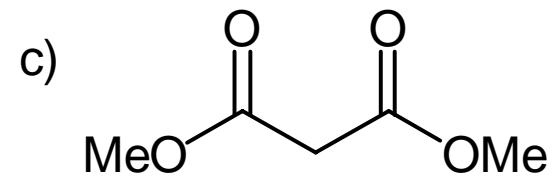
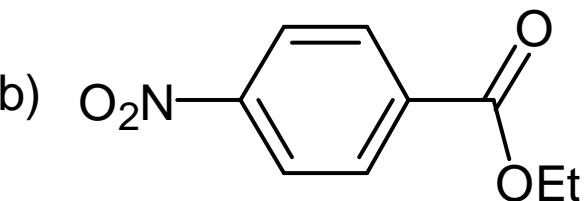
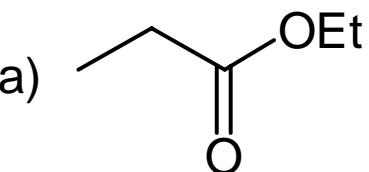


**Hydrogen bonding (red dots)  
between amide molecules**

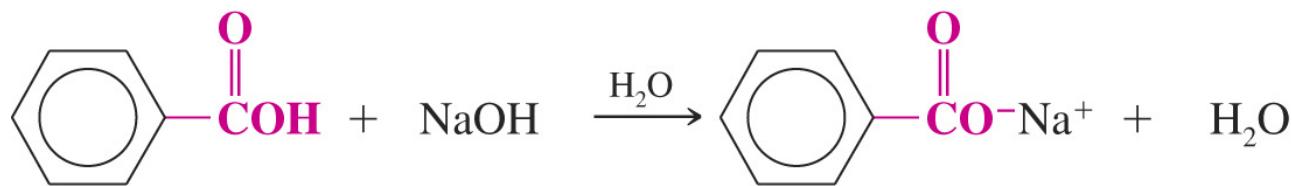


**Hydrogen bonding between amide groups of peptide  
chains. This interaction between chains (called a  
 $\beta$  sheet) is important to the structure of many proteins.**

Exercise 821:



## 2) 羧酸的酸性

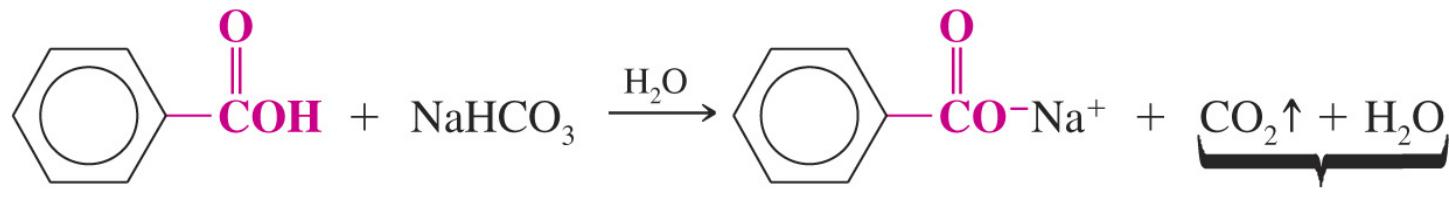


**Benzoic acid**  
(water insoluble)  
*Stronger acid*

*Stronger  
base*

**Sodium benzoate**  
(water soluble)  
*Weaker base*

*Weaker  
acid*

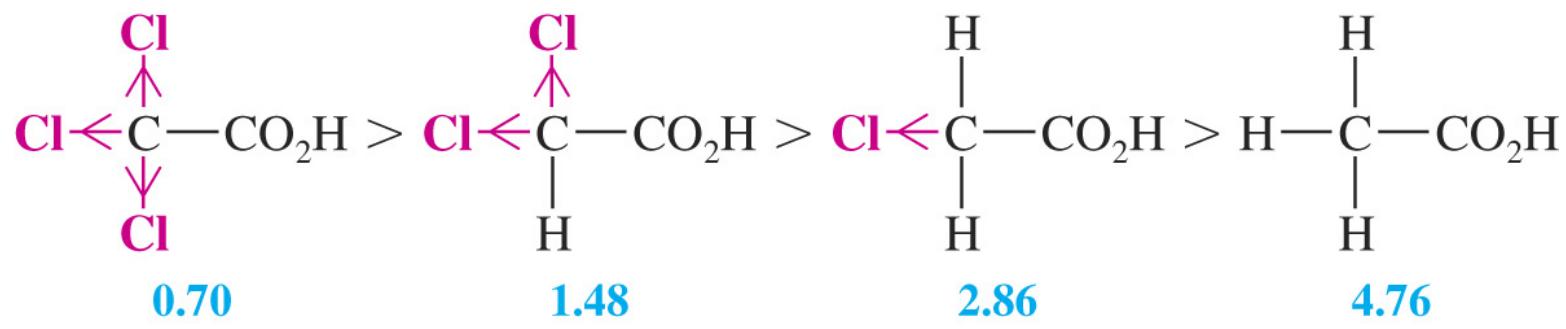


(water insoluble)  
*Stronger acid*

*Stronger  
base*

(water soluble)  
*Weaker base*

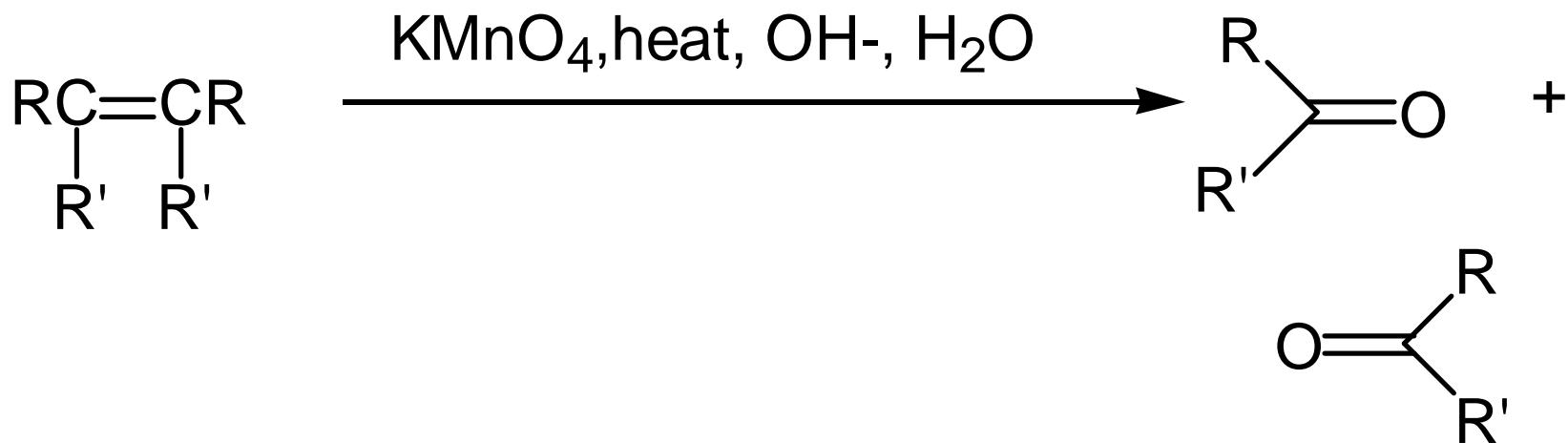
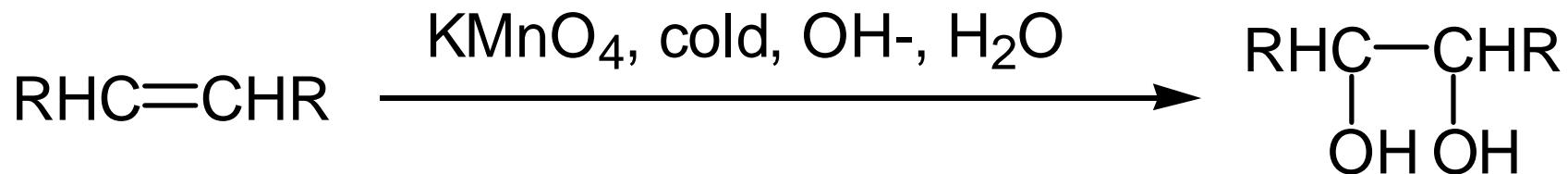
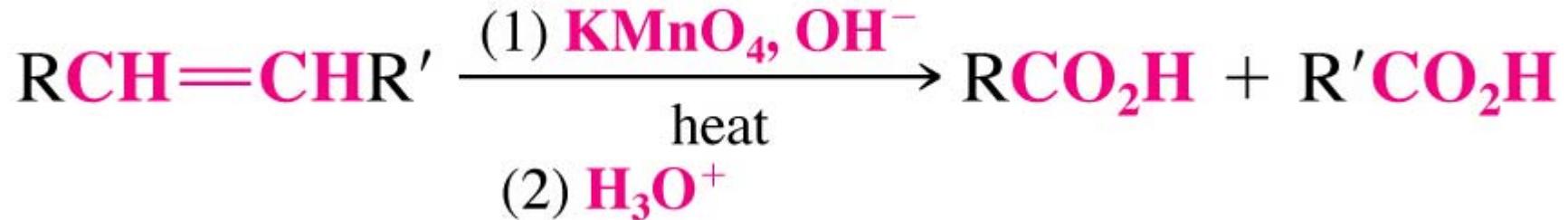
$\text{H}_2\text{CO}_3$   
*Weaker  
acid*

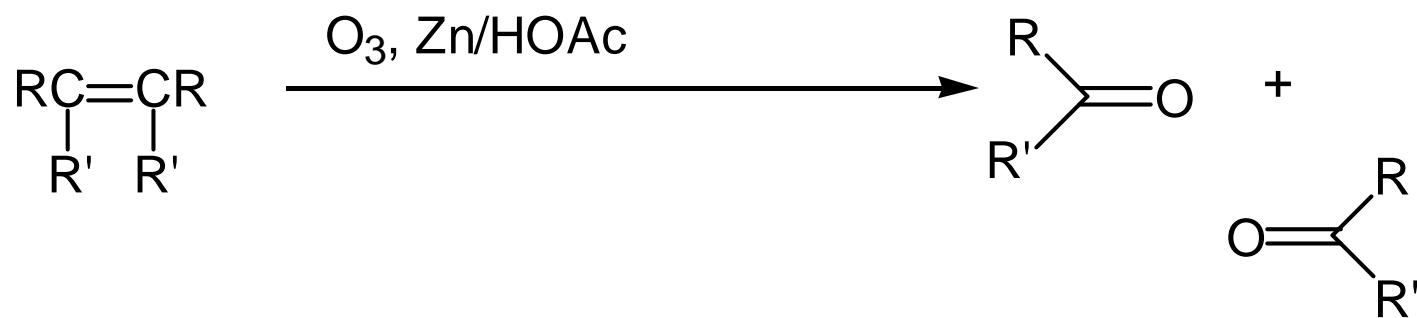
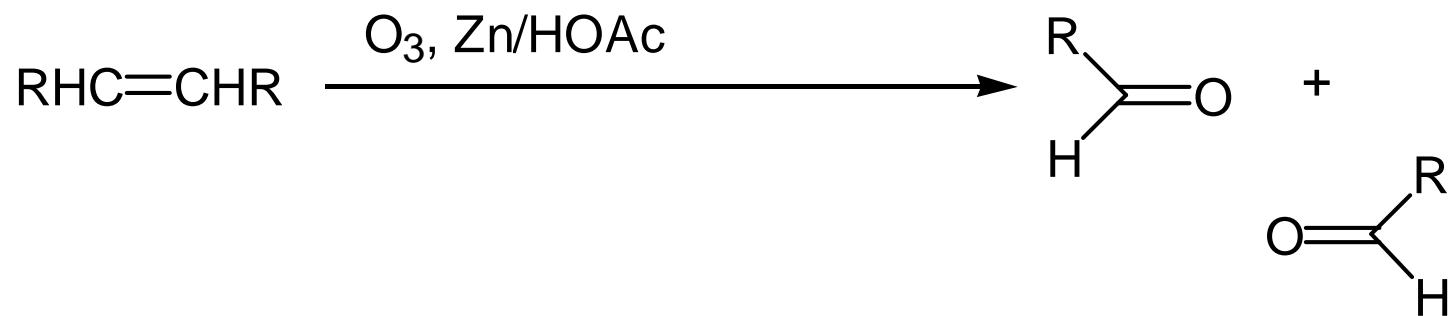
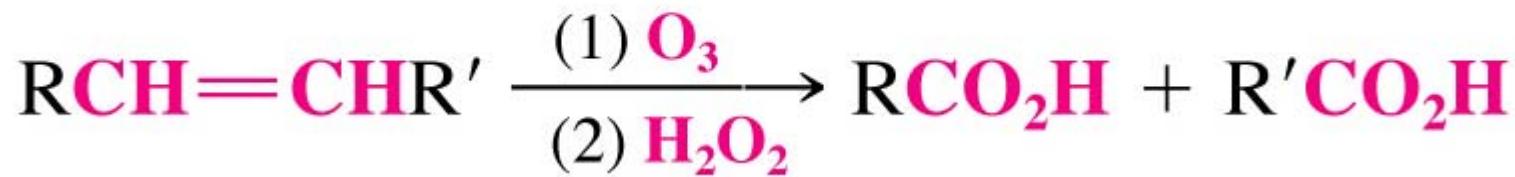


Exercise 818:

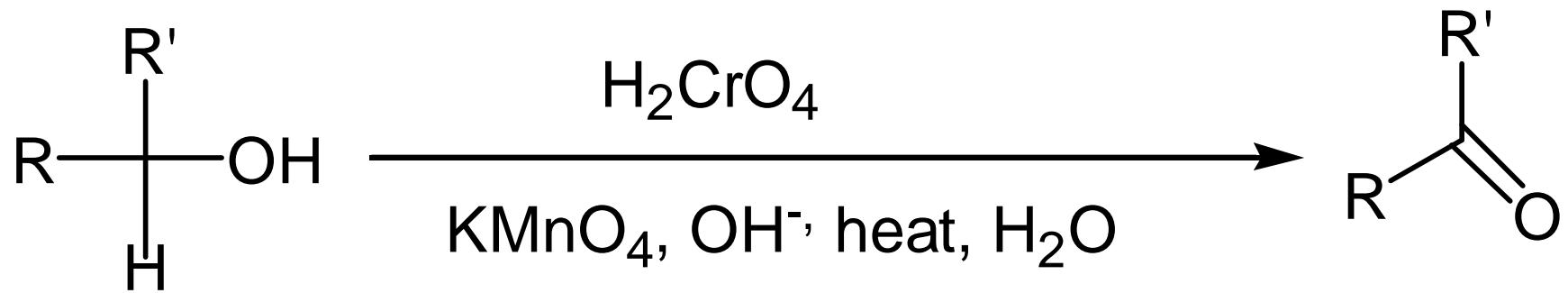
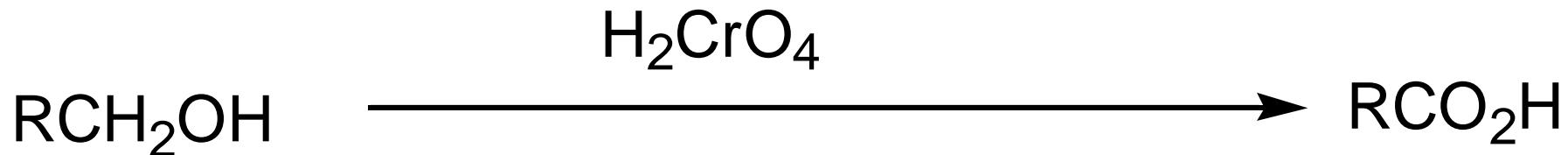
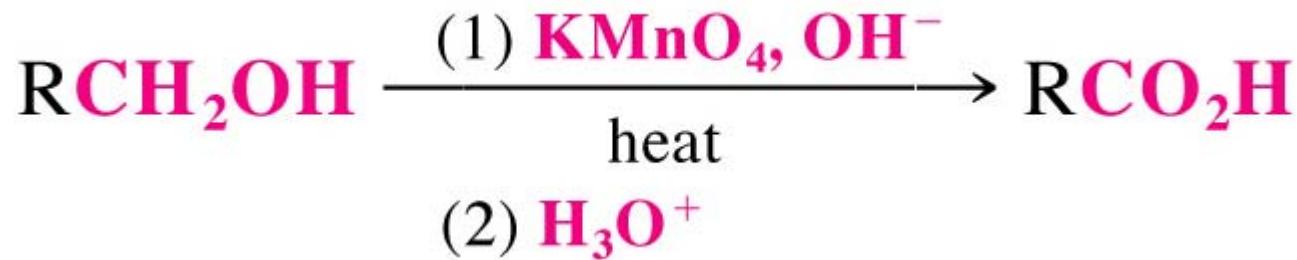
2) 羊酸的製備

a) 由烯烃氧化斷裂而制成

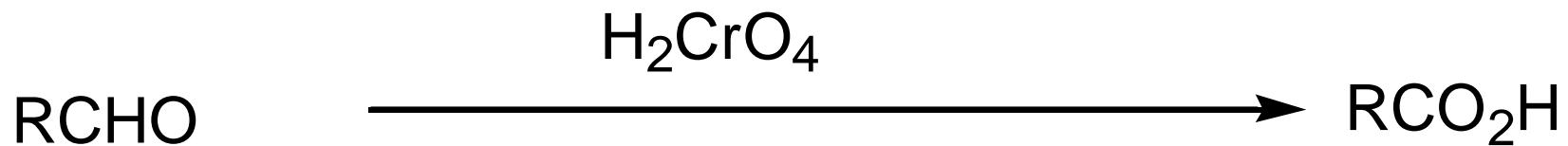




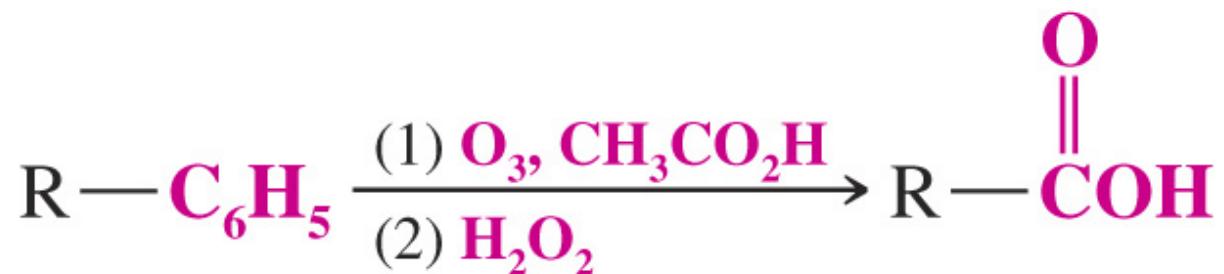
b) 由醇氧化而製成



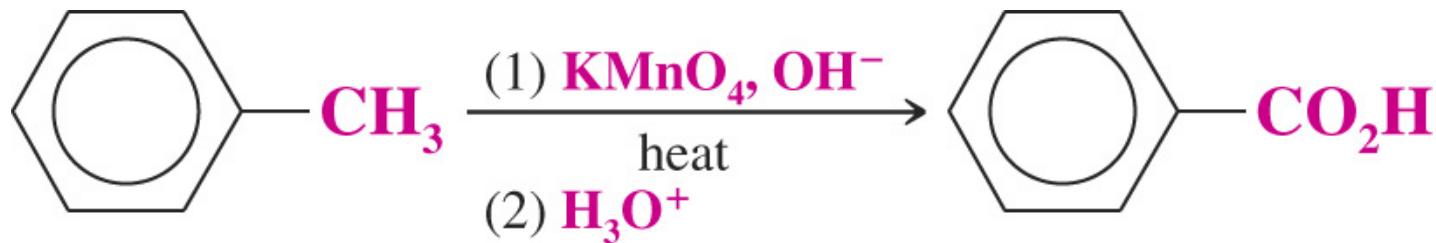
c) 由醛氧化而製成：



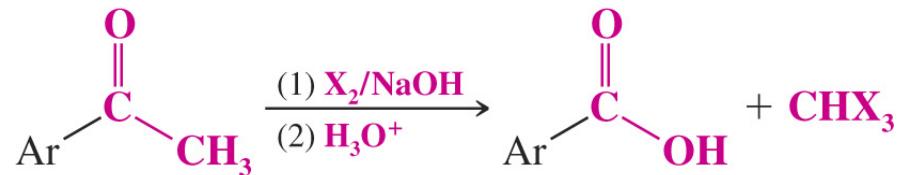
d) 苯环氧化



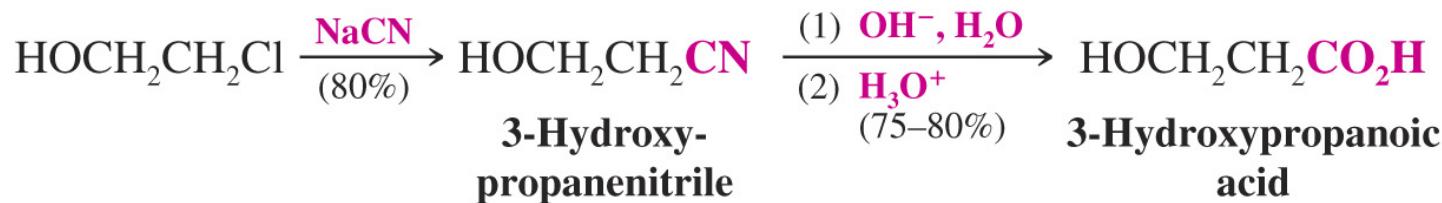
e) 烷基苯氧化



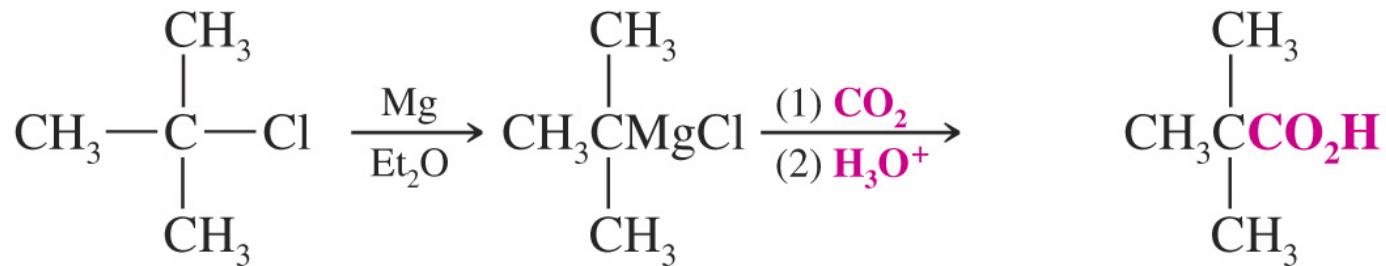
f) 甲基酮氧化(The Haloform Reaction): 解釋機制



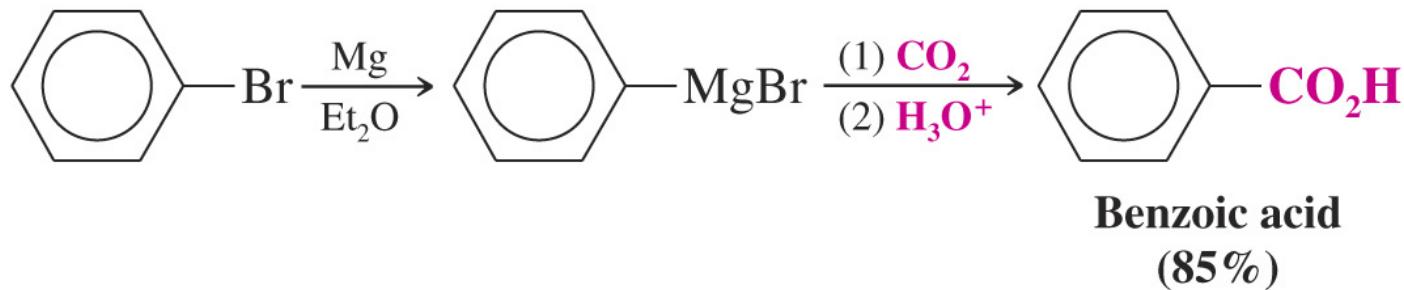
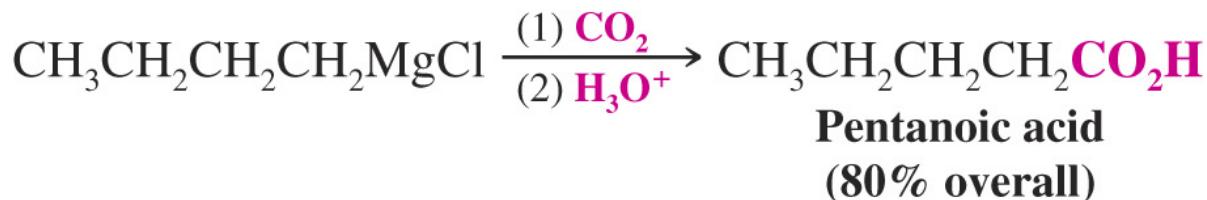
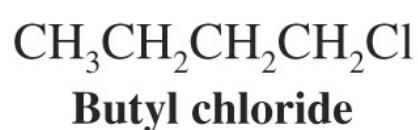
### g) Nitrile 水解



### h) Grinard反應

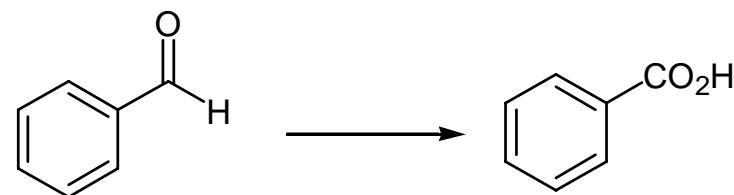
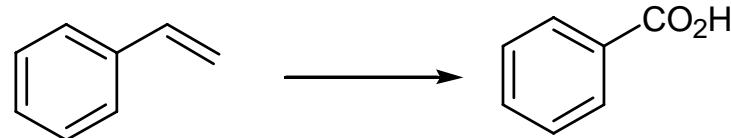
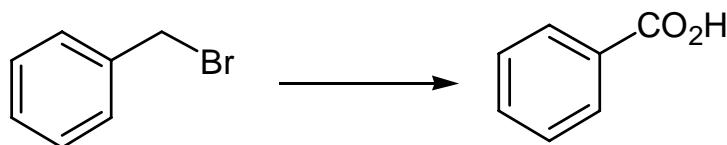
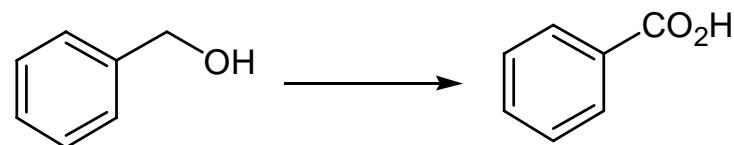
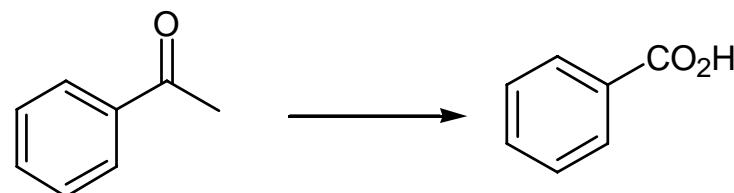


**2,2-Dimethylpropanoic acid  
(79–80% overall)**

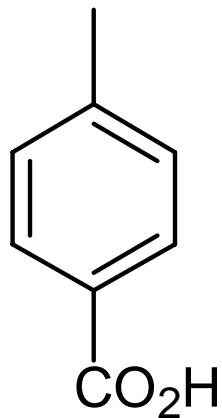
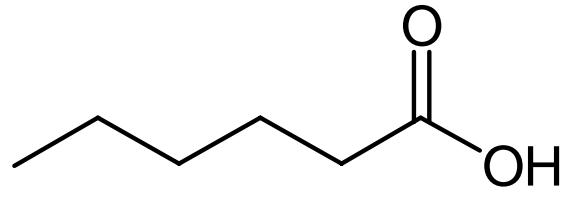
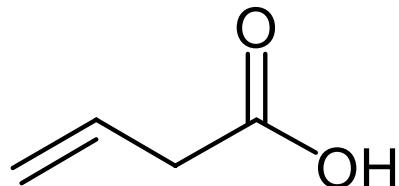
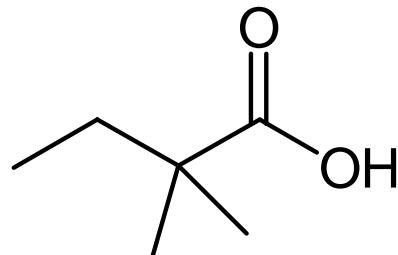
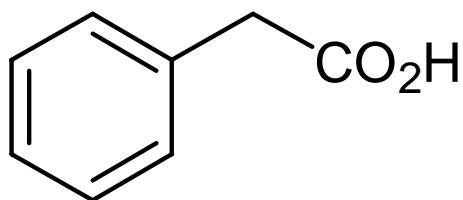


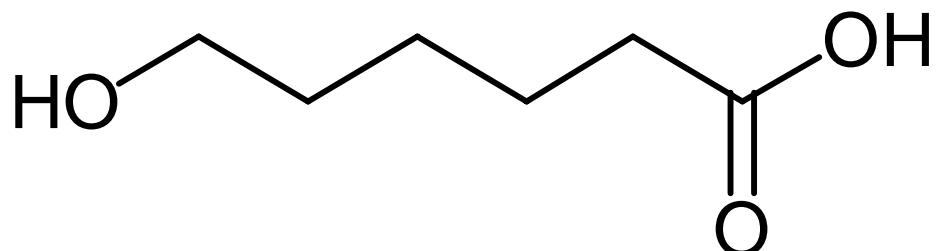
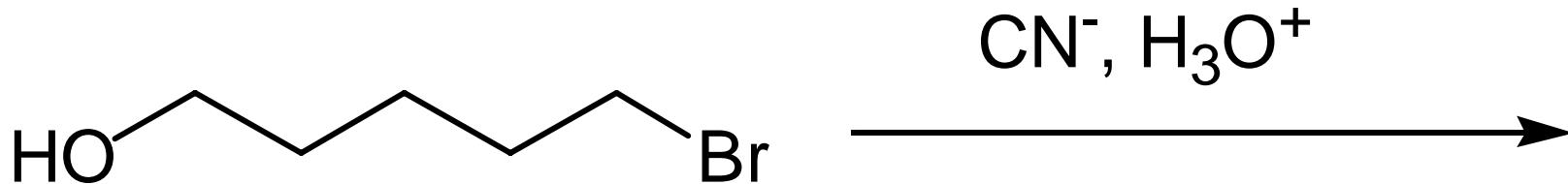
**Benzoic acid  
(85%)**

給出試劑即反應條件:



用Grinard反應合成下列化合物：





解釋

### 3) 羣酸衍生物的nucleophilic addition反應

